

# Lewis dot structures

1. In groups and on whiteboards, draw the Lewis Dot Structure for the following atoms and molecules (use notes and Periodic Table to help you)
  - a. O (oxygen)
  - b. Rb (rubidium)
  - c. CH<sub>4</sub> (methane)
  - d. CO<sub>2</sub> (carbon dioxide)

Get those down in your notebook...

For Lewis dot structures, the amount of dots around an atom equals the amount of valence electrons it has.